1250 Notes

Determination of Germanium in Some Plants and Animals

Sigeki Hara, Nanao Hayashi, Sigeo Hirano, Xi-Ning Zhong, Shigejiro Yasuda, and Hisashi Komae

Study of Environmental Science, Faculty of Integrated Arts and Sciences, Hiroshima University, Hiroshima 730, Japan

Z. Naturforsch. **45 c**, 1250–1251 (1990); received January 9, 1989/August 22, 1990

Germanium Determination, Hydride Generation, Inductively Coupled Plasma, Emission Spectrometry, Medicinal Plant

The Ge contents of plants and animals were investigated by a wet ashing procedure by hydride generation and inductively coupled plasma atomic emission spectrometry with flow injection. The analytical results obtained indicated that Ge contents widely vary in plant and animal kingdoms in the range of 8–203 ppb.

The role of Ge in biological processes is quite extensive, as it is an essential trace element. The occurrence of Ge in biomaterials is usually in the range of 0.1 to 1.0 ppm [1]. On the other hand, it has been reported that some medicinal plants contain large amounts of Ge, e.g. about 185 ppm for Zingibera rhizoma [2]. Studies indicated that rice tends to concentrate Ge in its shoots, and phytotoxic effects were observed [3]. Ge not only inhibited seed germination in certain strains of lettuce but also retarded seedling growth in several other plants [4].

Ge in biological materials is most often determined by one of two analytical methods: (1) spectrophotometry with phenylfluorone as color forming agent [5], (2) graphite furnance atomic absorption spectrometry (GFAAS) [6]. The method of Leek and Camphell [5] with phenylfluorone has such low sensitivity (detection limit, 0.5 ppm) that the values must be considered as unreliable. Dittrich *et al.* [7] demonstrated GFAAS for the determination of µg levels of Ge. The present paper deals with the determination of Ge in plants and animals by means of a very sensitive procedure *e.g.*, hydride generation and inductively coupled plasma atomic emission spectrometry (HGICPAES) [8] coupled flow injection. Of the

Reprint requests to Dr. N. Hayashi.

Verlag der Zeitschrift für Naturforschung, D-7400 Tübingen $0341-0382/90/1100-1250 \quad \$\ 01.30/0$

methods tested, wet digestion with H₃PO₄ and HNO₃ proved fastest, safest, and most efficient [9]. One gram of dried samples (dried 80 °C for 12 h) was weighed into a 100 ml tall beaker. After adding 30 ml of nitric acid and 1 ml of phosphoric acid, the samples were heated at 80 °C for 1 h covering with watch glass and allowed to stand overnight. After the predigestion, the samples were boiled at 200 °C until the dark HNO3 fumes had subsided. The resulting sample solution was concentrated to about 1-2 ml by evaporation and then analyzed by FIAHG and ICPAES. The standard conditions for FI and the instrumental conditions for ICPAES are summarized as follows: FI - NaBH₄ (5% m/V in 0.25% NaOH solution), buffer $(Na_2PO_4-H_3PO_4, 0.5 \text{ M}, pH 6.5)$, ICPAES – R.f. power (1.5 kW), outer gas (12 dm³ min⁻¹), intermediate gas (1.0 dm³ min⁻¹), carrier gas (0.3 dm³ min⁻¹), observation height (9 mm), wavelength (265.18 nm). As the method provides an absolute determination of Ge it has to be adjusted so that this amount falls into useful range of detection from about 1 ng to 10⁵ ng. The Ge is determined in an aqueous matrix at the part per trillion level by a combination of HG and ICPAES. The recovery of Ge was more than 95% even when 0.05 µg of Ge was added. The detection limit for Ge was 0.5 ppb (S/N = 3), which is 20 times lower to that of HG flame atomic absorption spectrometry. Standard deviation for a 10 ng/cm³ sample was 4% for 10 consecutive measurement. The plant and animal materials examined were collected in Hiroshima Prefecture in July 1987 except for the genera Miscanthus, Solidago, Artemisia, Oenothera, Polygonum, Chenopodium, and coal which were collected in Fukuoka Prefecture. 48 biological samples (plants and animals) were analyzed by this procedure. The results obtained are listed in Table I. Both plants and animals contained Ge at a ppb level. Park [4] reported the presence of Ge in the medicinal plant of a ppm level by means of flameless atomic absorption spectrometry. On the other hand, Shimomura [5] reported the presence of trace Ge lower than 3.7 ppb in the medicinal plants by atomic absorption spectrometry with electronthermal atomization. The present results show that medicinal plants contained only low amounts of Ge (12 to 37 ppb) and comparable with other plants. It was found that there is no relationship between their pharmacological effects



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.

Table I. Ge contents of the plants and animals.

Ge con	tent [ppb] in dry weight	Canoderma applanatum Microporus flabelliformis Coriolus versicolor	26 15 17	
Seed plant		Cryptoporus volvatus	n.d.	
Pinus densiflora (leaf)	23	Lichen		
Chamaecyparis obtusa (twig)	23	Parmeia tinctorum	38	
Miscanthus sinensis	20-55	Usnea diffracta	22	
Pleioblastus variegatus	33	Algae		
Phyllostachys niger	22	Ulva fasciata	n.d.	
Solidago virga-aurea (leaf)	11 - 14	Cladophora densa	62	
Solidago virga-aurea (root)	31 - 42	•	02	
Artemisia princeps	49	Medicinal plant	1.4	
Aralia cordata	11	Parantica japonica	14	
Eurya japonica (twig)	20	Houttuynia herna	20	
Chenopodium album	24	Swertia japonica	37	
Polygonum longisetum	44	Lycii fructus	13	
Rumex japonicus	8	Japanese geranium herb	13 17	
Rumex acetosa	13	Trapa japonica	1 /	
Denothera erythrosepala (stem)	n.d.	Animal		
Vicia angustifolia	8	Graptosaltria nigrofuscata	16	
Heterotropa takaoi	30	Cryptotympana japonensis	12	
Saxifraga stolonifera	22	Parantica sita	20	
Pteridophytes		Rat (internal organs)	22	
Pteridium aquilinum	n.d.	Domestic fowl (Gizzard)	10	
Equisetum arvense	20	Scomber scombus	19	
Lycopodium clavatum	27	Short-necked clam	18	
Osmunda japonica	20	Soil, coal and sea water		
Bryophytes		Soil	11-29	
Neckeropsis nitidula	66	Coal	1150	
Hypnum plumaeforme	105	Sea water	n.d.	
Rhacomitrium canescens	203	n.d., not detected.		
Thuidium kanedae	88	n.a., not actected.		
Chiloscyhus polyanthus	36			

Fungi

and the content of Ge in the medicinal plant. As various plants absorb Ge from soil, we examined Ge contents of soil to clarify the relationship between Ge contents of the soil and Ge contents of the plants. It was found that Ge contents of the plants did not depend upon that of the soil. All of the animals examined contained Ge in the range of 11 to 22 ppb. The sea water contained no detectable Ge in ppb level, while the coal concentrated Ge up to 1150 ppb. The largest value of Ge contents in

all the plants and animals was 203 ppb in the case of *Rhacomitrium canescens*. From these data, we assumed that Ge contents varies very widely in the plants and animal kingdoms in ppb level.

Acknowledgements

Grateful thanks are due to Prof. Kumamaru and Dr. Yamamoto (Department of Chemistry, Faculty of Science, Hiroshima University) for providing facilities of ICP and valuable suggestions.

- H. A. Schroeder and J. J. Balassa, J. Nutr. 97, 245 (1976); *ibid.*, J. Chron. Dis. 20, 211 (1967); H. A. Schroeder, M. Kanisawa, D. V. Frost, and M. Mitchener, J. Nutr. 98, 37 (1968).
- [2] N. H. Park, W. K. Lee, M. K. Park, and J. I. Park, J. Pharm. Soc. Korea 23, 141 (1979).
- [3] H. Matsumoto, S. Syo, and E. Takahashi, Soil Sci. Plant Nutr. (Tokyo) 21, 273 (1975); K. Tensho and K. L. Yeh, Soil Sci. Plant Nutr. (Tokyo) 22, 191 (1976); H. Matsumoto and E. Takahashi, Soil Sci. Plant Nutr. (Tokyo) 22, 191 (1976).
- [4] N. Sankhla and D. Sankhla, Naturwissenschaften 54, 621 (1967).

- [5] C. L. Like and M. E. Campbell, Anal. Chem. 28, 1273 (1956).
- [6] D. J. Johson, J. S. West, and R. M. Dagnoll, Anal. Chem. Acta 67, 79 (1973).
- [7] K. Dittrich, R. Mandry, W. Monthes, and J. G. Judelvic, Analyst 110, 169 (1985).
- [8] F. Nakata, H. Sunahara, H. Fujimoto, M. Yamamoto, and T. Kumamaru, J. Anal. At. Spectrom. 3, 579 (1988).
- [9] D. C. Reamer and C. Veillon, Anal. Chem. 53, 1192 (1981).

		×				
	Nachdruck – auch auszugs		er Genehmigung des Verla	ges gestattet		
Verantwortlich für den Inhalt: A. KLEMM Satz und Druck: Allgäuer Zeitungsverlag GmbH, Kempten						